

CONTACT INFORMATION

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PRINTED: 10-15-2012

ARIZONA DEPARTMENT OF MINES AND MINERAL RESOURCES AZMILS DATA

PRIMARY NAME: BUCK MOUNTAIN PROSPECT

ALTERNATE NAMES: HOBBLE MTN. GROUP

COCONINO COUNTY MILS NUMBER: 555

LOCATION: TOWNSHIP 25 N RANGE 3 E SECTION 36 QUARTER LATITUDE: N 35DEG 30MIN 01SEC LONGITUDE: W 112DEG 01MIN 42SEC TOPO MAP NAME: HOBBLE TANK - 7.5 MIN

CURRENT STATUS: OTHER

COMMODITY:

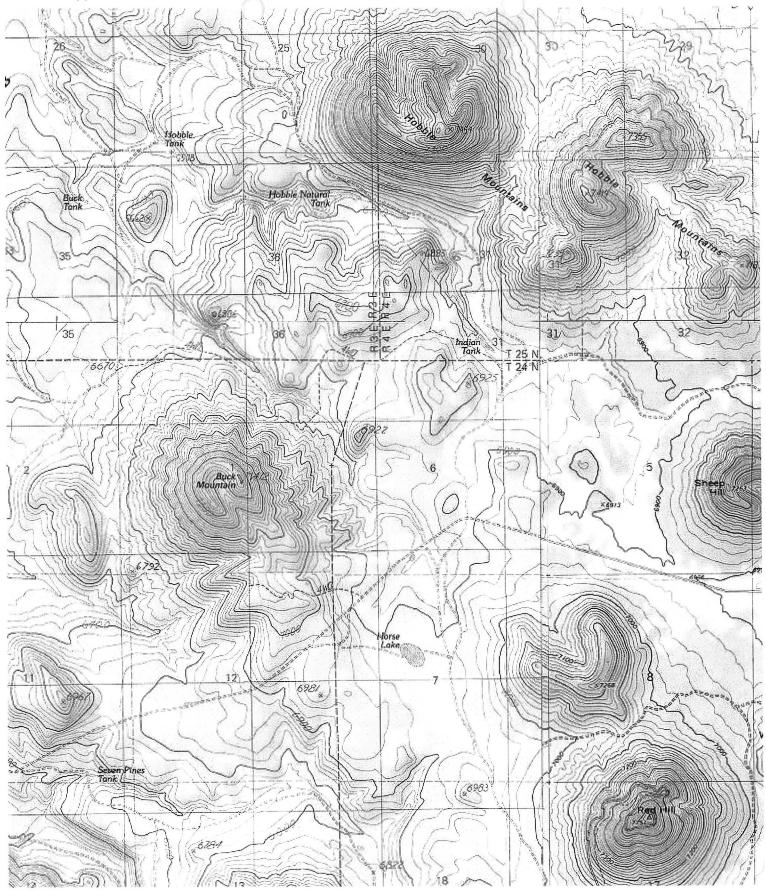
BIBLIOGRAPHY:

ADMMR BUCK MOUNTAIN PROSPECT FILE CLAIMS EXTEND INTO T24N R3E SEC 1, 2, 11, 12; T25N R3E SEC 23-27, 34-36; T25N R4E SECS 30 31; T24N R4E SECS 6, 7 Arizona Jepartment of Mines and Mineral mesources

VERBAL INFORMATION SUMMARY

May be Reproduced 1. Information from: Visct 2 AKA Williams Project Address: Tourstain Project 3. ADMMR Mine File Conterny F reste 2. Mine: Buck / 5. District 4. County: Coconino 6. Township <u>25N</u> Range <u>3E</u> Sec(s) <u>36</u> ing round an 7. Location: 8. No. of Claims - Patented 273 Unpatented 9. Owner (if different from above) Buck Mountain Propert 10. Address: 22.9 South Vover Delawave 19901 11. Operating Company: Represented by Alvin C. hason Ir. Ph 12. Pertinent People and/or Firm: Exploration Geochemist 1707 E. Weber Soite & Tempt AZ 13. Commodities: Precious Metals 14. Operational Status: Haw Prospect 15. Summary of information received, comments, etc.: On March 7. 1989 (at the mulation of Gutierrez. with John Gutierrez; Kobin AZ Zone Office, US Forest Service Imet USFSALBUQUES TOM Gillett. Struthy AZ Zoneo Hice: Doger Marion Ken edu o Mines salo 10000 Gary Slusher and Mike Johnson Genter and Mountain Frequesties, Inc nson representing Buck NO C. the Forest Service in the vicinity of Sections 20 previously by Dr. Johnson as Indicated on map bs (includining tsent =plit to separate Ye h took separa Bof M Reno Researce Centers Samples had there assayed from these tocations und sonking Assay Stree. 193 at 1/4. The same group met proceedure 221 a om man to the assay results auro Gred Sprenger Bondar Clega Inc head and along with a report ka. BYDGEED Date: 6/30/89 (Signature) ADMMR

ACME Mapper 2.0 - 10.2 mi NW of Kendrick Peak AZ

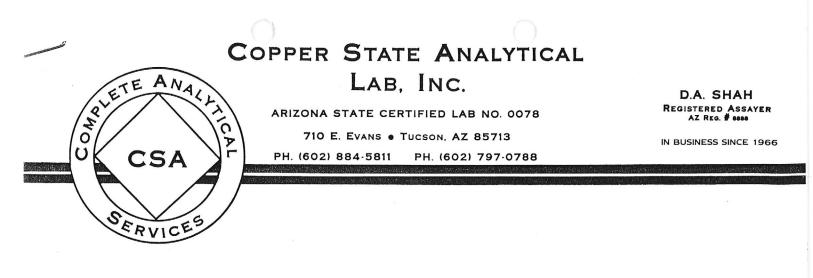


Buck Mountain Prospects

BUCK MIN PROPERTIES Mar. 7 89 SAMPLING John Gatierrez, USTS AZ ZONE Ken Broachead USBM Reno NU Mike Johnson TEMPA AZ RESEARCH Dick Beard AZ Dept. of Mirres Phoenix AZ Gary Slusher St. Land Dept. Phoenix, Az Al Johnson Tempe, Az Regearch ROGER MARION ALBUQ, NM. USFS Tom Gillett USFS Karlach, Wms. Robin Strathy USFS AZ Zong Phx

On Murch 7, 1989 at the invitation of John Gutierrez, AZ Zone Office of the D.S. Forest Service, I met with him; Robin Strathy of his office; Royer Marion, USFS Albuquerque, NM; Tom Gillett, USFS Karbab, W.H. Jones; Ken Broadhead, US Bureau of Mines, Keno Research Contes; Gary Slusher, AZ Statehand Dept ;; Mike Schoson and Dr. Albin C. Johnson of Tempe, After meeting at the Karbub Forest Supervisors Office in Willounds we proceeded to the Bock Mountain Aning Properties, Inc. I CN DU IT X Pauticipents Sample Nos. # 1 Purpose 28057 Location HZ. 28058 #3 Discossion 28059 #4 Asserpting 28060 Assing Ross 145 #5 28061 Alvine Statison -#6 28062 Ant Hill Ken Brondhard Auntihill Both des Llegg Skylme Inon King C De Don No Conclusions. ISCAM

3 Mr



Mr. Alvin Johnson	Job: 19346
1707 East Weber Drive, Suite #8	Received: 03/10/89
Tempe, Arizona 85281	Reported: 03/15/89
	Sample No.: 7
	Elements: 2

Project:-William	n, AZ		
Elements Units	Au OPT	Ag OPT	
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0.8mg - BM-4 1.6mg - BM-2 2.2mg - BM-1 20.6mg - BM-5 24.4mg - BM-3 46.5mg - BM-1 88.3mg - BM-6	0.024 0.016 0.90	1.342.964.6037.5846.4891.90173.32	

NOTE:-Calculations are based on 15gm wt of sample. Client provided the dore from fire assay.

MAR 2 2 1989 Юц. (_{— - -} MINESAL RESUL ...



ALVIN C. JOHNSON, JR., PH.D. EXPLORATION GEOCHEMIST 1707 EAST WEBER DRIVE, SUITE 8 TEMPE, ARIZONA 85281

3/13/89

Mr. Richard Beard Mining Engineer Arizona Department of Mines and Mineral Resources Mineral Building Fairgrounds Phoenix, AZ 85007

Dear Dick:

Enclosed please find a set of copies of the cupelled prills that we obtained from the fire assay demonstration that was performed for the U.S. Forestry last week. These prills were sent to Mr. D. A. Shah of Copper State Analytical for analysis and according to him the results should be ready this Wednesday.

Also enclosed is a copy of the assay procedure that was used during the above demonstration. As you will note, this fire assay method is rather standard except for the demanded use of an electric furnace and for the two sintering steps. Since the ores similar to the Williams ore typically have relatively large amounts of such elements as As, S, Sb, Hg, T1, etc. this fire assay procedure also includes two sintering steps that we have found absolutely necessary for completion of a successful fire assay on this material.

When Mr. Shah completes his analysis of the above prills I will foreward you a copy of the results.

Sincerely,

Alvin C. Johnson, Jr., Ph.D.





R . - - - 13

ALVIN C. JOHNSON, JR., PH.D. EXPLORATION GEOCHEMIST 1707 EAST WEBER DRIVE, SUITE 8 TEMPE, ARIZONA 85281

PROCEDURE FOR THE ANALYTICAL TESTING OF AU AND AG IN GEOLOGIC MATERIALS

- Note: The following analytical procedure has been developed as a preliminary fire assay technique for certain refractory ores, specifically for basalt and related rock types. The furnaces that were used in this study were electric and the fluxes that are used in the following procedure were developed for the internal atmosphere of an electric furnace. In a relative sense, the internal atmosphere of an electric furnace might be considered as reducing and that of an air-blown, muffle-type furnace as oxidative. IT WOULD SEEM IMPORTANT, THEREFORE, THAT IN THE FOLLOWING PROCEDURE AN ELECTRIC FURNACE BE USED.
- A. Matrix preparation.
 - 1. Grind 15 grams of the sample to a fine powder (±100 mesh).
 - 2. Mix thoroughly with 5 grams of Na_2O_2 using a mortor and pestle.
 - 3. Place in a clay crucible and sinter in furnace at ±850°C for 1 hour. Rabble this sinter periodically.
 - Remove the sinter from the furnace and after cooling somewhat remove the sinter from the crucible. After the sinter has cooled to room temperature grind it to a fine powder.
 - 5. Place the Na₂O₂ sinter in a mortor and pestle and add 10 grams of litharge. Mix thoroughly and replace in the same crucible.
 - 6. Roast in furnace at ±850°C for 1 hour.
 - 7. Sinter step #A-4.
 - 8. Place the sinter in a mortor and pestle and add the following flux:
 - a. 40 gram litharge.
 - b. 30 gram sodium carbonate.
 - c. 30 gram borax.
 - d. 5 gram SiO₂.
 - e. 5 gram flour.

Mix thoroughly and replace in the same clay crucible.

- 9. Fuse at 1000°C for 45 to 60 minutes (or until the fusion becomes quiescent).
- 10. Pour in cast iron mold. Clean slag from the lead button. The lead button should weigh approximately 40 grams.
- B. Scorification of lead button.
 - Place lead button in scorifying dish and add approximately 5 gram of assay lead. Cover the lead with borax and place in furnace at 1000°C. Scorify the lead until approximately one-half of the original lead remains. This step will take approximately 1.5 hours.
 - 2. Clean the resulting lead button of slag and weigh. If the button does not weigh approximately 25 grams add sufficient assay lead to make up the difference.

C. Cupellation.

- Prior to actual cupellation place the cupel in the furnace for ± 15 minutes in order to remove any water that may be sorbed in the cupel.
- 2. Place the lead button (and any makeup assay lead that might be necessary) in the cupel at a furnace temperature of 1000°C. Except for inspection, keep furnace door closed during the cupellation process.
- 3. Remove the cupel from the furnace and let it cool. Remove the prill and carefully clean it. Weigh and record.

The prill resulting from the above fire assay procedure may be analyzed for Au and Ag by normal parting and weighing procedures or by atomic absorption. U.S. Forest Service Sample BM-1 Soil Dore wt for 15 gm sample = 46.5mg

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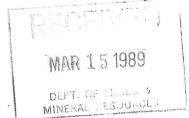
Williams, Az. Project

U.S. Forest Service Sample BM-1 Basalt Dore wt for 15 gm sample = 2.2mg

Williams, Az. Project

U.S. Forest Service Sample BM-2 Basalt Dore wt for 15 gm sample = 1.6 gm mg

Williams, Az. Project



U.S. Forest Service Sample BM-3 Basalt Dore wt for 15 gm sample = 24.4 mg

Williams, Az. Project

U.S. Forest Service Sample BM-4 Basalt + soil Dore wt for 15 gm sample = 0.8 mg

Williams, Az. Project

"lost"

U.S. Forest Service Sample BM-5 Soil Dore wt for 15 gm sample = 0.4 mg

Williams, Az. Project

U.S. Forest Service Sample BM-6 Basalt Dore wt for 15 gm sample = 88.3 mg

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Williams, Az. Project

U.S. Forest Service Sample BM-5 Basalt Dore wt for 15 gm sample = 20.6 mg

Williams, Az. Project



ALVIN C. JOHNSON, JR., PH.D. EXPLORATION GEOCHEMIST 1707 EAST WEBER DRIVE, SUITE 8 TEMPE, ARIZONA 85281

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The prill resulting from the above fire assay procedure may be analyzed for Au and Ag by normal parting and weighing procedures or by atomic absorption.



United States Department of the Interior

BUREAU OF MINES

RENO RESEARCH CENTER

1605 EVANS AVENUE RENO, NEVADA 89512-2295

May 18, 1989

John Gutierrez U.S. Forest Service 2324 East McDowell Road Phoenix, Arizona 85006-2497



Dear John,

Enclosed are the analytical reports on the rock and soil samples taken March 7, 1989, at the so called Buck Mountain Mining Project. The analyses are divided into two separate types, one to generally categorize and identify the material, and the other to quantify the precious metals content. The categorization tests as shown in exhibits 5, 6, and 7, show the volcanic cinders and soil to be primarily quartz, pyroxene and feldspars. There were no concentrations of elements present which would indicate that the samples would present any particular problems to standard fire assay techniques.

Exhibits 1 and 2 are fire assay results on my original splits for gold and silver. Exhibit 2 also shows the results following the procedure set forth by Alvin Johnson. A copy of this procedure is also enclosed as exhibit 8. Exhibit 3 shows regular fire assay results on your portions of material which I had re-requested from you and which had been forwarded to me by Bondar-Clegg. Exhibit 4 shows the results from a combined fire assay/ICP analysis for the platinum group metals.

Comments on exhibit 8: Dr. Johnson's assay procedure.

General: The procedure is unduly long, time consuming and completely unnecessary.

Specific: Steps 2 through 4: There is no reason for adding and sintering with sodium peroxide. There are no minerals present, or carbon or sulfur, which would require such an oxidizing agent.

Steps 5 through 7: No reason for the addition and sintering with PbO (litharge). I can not see the rational for such a step, particulary after the preceeding peroxide fusion.

J. Gutierrez

Steps 8 through 13. Scorification. There is no reason for the use of the excess flour to produce a large lead button which must be later scorified. Scorification of lead buttons is used primarily to either reduce the size of the lead buttons or to remove impurities such as copper and nickel from the buttons so that they will cupel properly. The elements however are not present in this particular material. Therefore, the whole exercise of excess flour and scorification serves no useful purpose. However, it may be detrimental to the procedure since the use of excess flour creates a lead deficient slag which may cause problems.

C. Cupellation 2. The door cannot be completely closed so as to exclude air since the air or actually oxygen is a necessary ingredient in the process. Without oxygen the molten lead will not form litharge (PbO) which will be adsorbed into the cupel--the purpose of the process. The cupellation temperature of 1,000° C is excessive and will lead to silver losses.

Generally speaking values obtained by using Dr. Johnson's procedure showed no difference from those obtained from normal fire assay techniques. The results of our work showed no large amounts of either gold or silver. Sample BM-4, however, did show low but anomalous amounts of gold.

Sincerely,

emeth S. Sree Kenneth G. Broadhead Research Supervisor

Enclosures

cc: Dr. Al Johnson 1707 East Weber Drive Suite 8 Tempe, Arizona 85281

> Dick Beard Department of Mines and Mineral Resources Mineral Building-State Fair Grounds Phoenix, Arizona 85007

2

SAMPLE ANALYSIS REPORT

Samples submitted by

BROADHEAD 67.171

Date of Request Date Received by Fire Assay 2 189 Month Day

17:31 \bigcirc

Year

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SAMPLE ANALYSIS REPORT

Samples submitted by

BROADHEAD GZ-172

Date of Request Date Received by Fire Assay 61 Month Day

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Year

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1/2 A.T. samples

SAMPLE ANALYSIS REPORT - X'ray Diffraction

3

Joys

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177	m	m	Poss-t		1		t			
	m	M			. m			m		
17	÷	m-t	Poss-t			Poss-t				
		'			•	1				
4 Flow	reseace									
A DESCRIPTION OF THE REAL PROPERTY OF THE PARTY OF THE PA	A COLORADO C	lt. e	st. o	f ele	men	As 2:	12			
AL	si	ic	Ca	τ:	Cr	Mn	Fe	54	21	Ba
9.	32,	а.	3,	. 6	.03	.)	5.	.07	,04	,06
9.	29.	2,	۲.	.7	.04	. 1	6.	.1 .	,03	.07
.10.	3D,	3.	4.	.6	.02	.09	5.	,09	,03	.07
12.	24.	. 9	6.	1.	103	,]	10.	,1	.02	,06
10,	28,	2.	6.	.8	.03	,1	7.	.08	103	,06
<i>q</i> .	31.	2.	Ч.	.7	.03	.1	5.	.08	.03	,05
10.	23.	1.	·7.	2.	107	-1	1	.1	,03	.04
9.	31.		3.							:05
Q+2 =	Quartz			Amp=	Amph	libole		und =	unden	+
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Remarks:

A. g/l Reported as B. μ g/ml D. ppm element unless otherwise noted. C. & E. oz/ton oxide M = major prob = probable m = minor Analyst: KB Date: 3/23/89t = trace poss=possible × - , *

INDUCTIVELY COUPLED PLASMA ANALYSIS

SUBMITTED BY: SAMPLE SET NO				\$ M.W.	[CKES				3/13/89 3/20/89
SAMPLE		AG	AL	AS	E:	ĥ	1. 1. 1. 1.	Ë I	CA
BW-1		26	6.8% *				0.2		
-2		30	7.6% *	120			0 * 5		
·····		37	8.1% *		0.23	3% <	0.2	230	5,5%
4	炏		9,5% X		0.10		0.2	* 120	6.0%
- 5	Ж.		7.6% *		0.1		0.2	190	
···· É.		20	6.7% *				0.2	170	
-7	炏	16							
	×	17	6.3% *	63	0.1	4% <	0.2	* 130	2,2%
SAMPLE		CD							
№M-1		8.7						2.57	
-2		9.0	42	350			4,7%	2,77	120
÷			37			3	4.5%	3.67	4 120
4		6.5	48	75	4	1	7.7%	0.757	4 53
5		9.1	43	220	61	3	5.6%		100
-6		6.5	38	320	53	X.	4.5%		6 95
-7		8.2	64	350	8	9	9.2%	1 - 23	(58
-8		6.0	46	320	4	3	4,3%	2,21	. 76
SAMPLE									
BM-1			2 , 2%			3			97
2		23				7			110
Ξ-		25							92
A									51
		33				2			93
-6			3.0%			3		51	
		6.7							
-8		31	2.1%	0.19%	× 1	5	1.4%	37	95

RESULTS ARE REPORTED IN PPM UNLESS OTHERWISE INDICATED (UG/ML=MICROGRAMS/ML; G-GRAMS/L)

NOTE: < INDICATES THAT THE RESULT IS LESS THAN THE GIVEN VALUE * INDICATES THAT THE RESULT IS NEAR THE DETECTION LIMIT AND MUST BE INTERPRETED ACCORDINGLY 10-

INDUCTIVELY COUPLED PLASMA ANALYSIS

SUBMITTED BY: SAMPLE SET NO.		OADHEAD GZ-171	ANALY	ST:	M,WI					3/13/89 3/20/89
SAMPLE		P	PB		SB	SN		SR	TI	V
BM-1		0.17%	130	*	61	44		480	0.46%	160
2		0.25%	140	*	73	61		650	0.53%	180
		0.27%	130	<	60	58		520	0.49%	170
<i>Q</i>		0.10%	73	\leq	60	24		660	0.99%	260
- 5		0.24%	130	*	65	40		560	0.64%	180
-6		0.26%	92	<	60	27	i.	520	0.51%	170
7		0.24%	79		60	21	(540	1.3%	270
- 9		0.13%	90	<	60	18	2	290	0,48%	150
SAMPLE		Ŵ	Y		ΖN	ZR				
BM-1	*	120	26	4144 AUCE 2014 AU	270	140	anda			
-2	*	140	-7) E5		260	190				
		150	26		380	180				
4	*	120	26		180	120				
-5	*	150	28		280	140				
-6	ж	110	27		170	140				
7	ж.	120	27		190	200				
8	漱	110	26		120	110				

RESULTS ARE REPORTED IN PPM UNLESS OTHERWISE INDICATED (UG/ML=MICROGRAMS/ML; G=GRAMS/L)

NOTE: < INDICATES THAT THE RESULT IS LESS THAN THE GIVEN VALUE * INDICATES THAT THE RESULT IS NEAR THE DETECTION LIMIT AND MUST BE INTERPRETED ACCORDINGLY

CARBON SULFUR ANALYSIS REPORT

Samples submitted by BROADHEAD Request Date 3 / 13 / 89 Type of Sample <u>67.171</u>

14:28 03/	16/89			
ID CODE	WEIGHT	CARBON%	SULFURX	
C17100001			0.0218	
C17100002		0,7008		
C17100003		0.1770	0.0294	
C17100004		0.7789	0.0183	
C17100005		i.138	0.0384	
C17100006		0.7652	9.0327	
C17100007		0.1055	0.0213	
	0.0301	1.176	0.0197	
C SD { C000000004 C000000003	0.0301	0.9467	0.0227 [0.9500 ± 0.053
	0.0001	0.9533 0.2974	0.0246)	
5 500 {C000000002 (C000000001	0.0000 0 0700	0.2774 0.2906	2,86	2.8±1%RSD
C	010000	V. 1700	∠.(4)	

Remarks TOTAL CARbon- Sulfur

· · · · · · · ·

Analyst F. Godse

Date Completed 3/16/89

7



ALVIN C. JOHNSON, JR., PH.D. EXPLORATION GEOCHEMIST 1707 EAST WEBER DRIVE, SUITE 8 TEMPE, ARIZONA 85281

PROCEDURE FOR THE ANALYTICAL TESTING OF AU AND AG IN GEOLOGIC MATERIALS

Note: The following analytical procedure has been developed as a preliminary fire assay technique for certain refractory ores, specifically for basalt and related rock types. The furnaces that were used in this study were electric and the fluxes that are used in the following procedure were developed for the internal atmosphere of an electric furnace. In a relative sense, the internal atmosphere of an electric furnace might be considered as reducing and that of an air-blown, muffle-type furnace as oxidative. IT WOULD SEEM IMPORTANT, THEREFORE, THAT IN THE FOLLOWING PROCEDURE AN ELECTRIC FURNACE BE USED.

A. Matrix preparation.

- 1. Grind 15 grams of the sample to a fine powder (±100 mesh).
- 2. Mix thoroughly with 5 grams of Na_2O_2 using a mortor and pestle.
- 3. Place in a clay crucible and sinter in furnace at ±850°C for 1 hour. Rabble this sinter periodically.
- Remove the sinter from the furnace and after cooling somewhat remove the sinter from the crucible. After the sinter has cooled to room temperature grind it to a fine powder.
- 5. Place the Na_2O_2 sinter in a mortor and pestle and add 10 grams of litharge. Mix thoroughly and replace in the same crucible.
- 6. Roast in furnace at ±850°C for 1 hour.
- 7. Sinter step #A-4.
- 8. Place the sinter in a mortor and pestle and add the following flux:
 - a. 40 gram litharge.
 - b. 30 gram sodium carbonate.
 - c. 30 gram borax.
 - d. 5 gram SiO₂.
 - e. 5 gram flour.

Mix thoroughly and replace in the same clay crucible.

- 9. Fuse at 1000°C for 45 to 60 minutes (or until the fusion becomes quiescent).
- 10. Pour in cast iron mold. Clean slag from the lead button. The lead button should weigh approximately 40 grams.
- B. Scorification of lead button.
 - Place lead button in scorifying dish and add approximately 5 gram of assay lead. Cover the lead with borax and place in furnace at 1000°C. Scorify the lead until approximately one-half of the original lead remains. This step will take approximately 1.5 hours.
 - 2. Clean the resulting lead button of slag and weigh. If the button does not weigh approximately 25 grams add sufficient assay lead to make up the difference.
- C. Cupellation.
 - Prior to actual cupellation place the cupel in the furnace for ± 15 minutes in order to remove any water that may be sorbed in the cupel.
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 - 3. Remove the cupel from the furnace and let it cool. Remove the prill and carefully clean it. Weigh and record.

The prill resulting from the above fire assay procedure may be analyzed for Au and Ag by normal parting and weighing procedures or by atomic absorption.



United States Department of Agriculture Forest Service Arizona Zone Office 2324 E. McDowell Road Phoenix, AZ 85006 602 225-5261

Reply To:2810

Date: June 28, 1989

Subject: Buck Mtn Sampling Map

To: T.Gillett, Kaibab NF

Attached you will find a map of the sample locations for the assaying work conducted recently near Hobble and Buck Mountain. Also included is Al Johnson's previous sample locations & results. This information should be kept in your file for possible future reference.

JOAN GUTTERREZ Mining Engineer

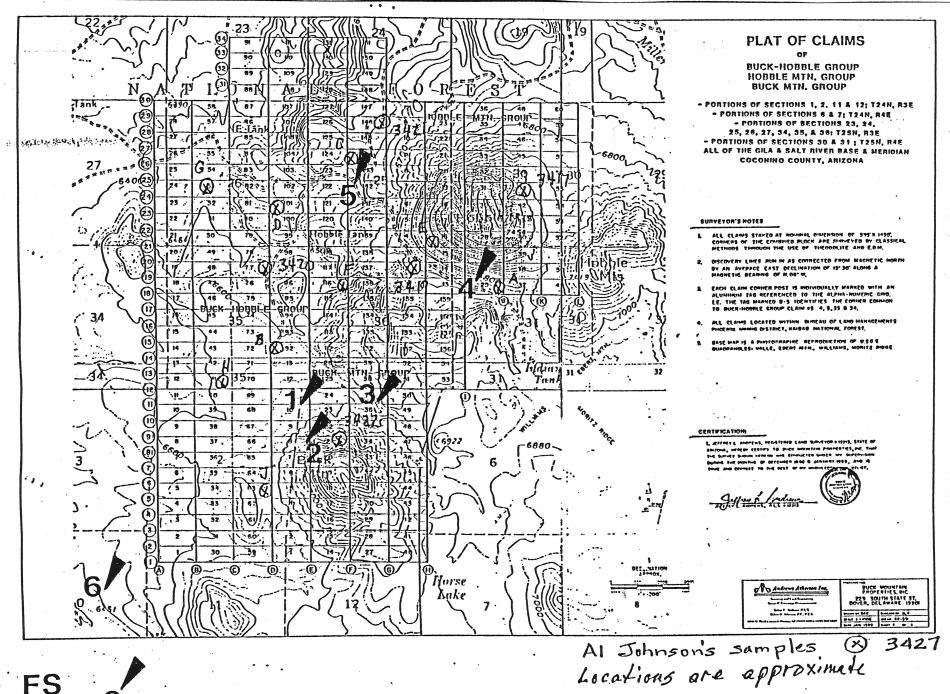
cc: R.Marion, Regional Office K.Broadhead, NV Bureau of Mines

- D.Beard, AZ Department of Mines & Mineral Resources Gary Slusher, AZ Department of State Land & Mineral Resources

RECEIVED JUN 2 9 1989 DEPT. OF ME . MINERAL RESOURCE

Caring for the Land and Serving People





FS

2. 5. 	Assay Sum	mary Sheet	(Fire Assy -	Williams Project 1/2 Assay Ton)
Samply ID	Dorie WH+ (mg)			Au (Troy 02/ton)
A	6.Z.	0. /	12,2	0.2
	8.9	0.3	17.2	9.4
	18.2	0.2	32.4	0.4
В	9.8	0.4	18,8	۵.9
	11. 3	0./	22,4	0.2
	8.7	0.2	17.0	0,4
C	22.2	0,4	43.6	D.8
	10.9	0./	21,6	0.2
	16.1	0,1	32,0	0,2
D	10,2	0,2	20.0	0.4
	26.7	0.Z	53,0	0.4
	25,4	۵./	50.6	0,2
E	18.1	5.2	35,8	0,14
	12.6	0,2	24,8	0,4
	16,0	0./	31,8	0.2
F	20.9	0, 3	41,Z	0,6
	18.1	0,2	35.8	0.4
	15,4	0,2	30,4	0,4
6	8.Z	0,1	16,2	0,2
	9, 5	0.2	18,6	0.4
	11.4-	0,/	22,6	0.2
4	24.7	0,2	49,0	0,4
	21.9	0.Z.	43.4	0,4
-	18.0	0,2	35,6	0,44
\mathcal{T}	15. Z	0,1	30,2	0,Z /
	11,9	0.3	23.2	0.6
	12.0	. 0,4	23.2	0,8
3417	.32.2	0.Z	44,0	0.4
	25,2	0,/	50, Z	0.2
11	26,1	0.2	51.8	0.4
3419	+11.07.5	0,136	21,778	0.272
	.9,2	0.1	18.2	0.2
	20,1	0,2	39.8	0,4
	11.8	0,/	23,4	0.2
	- * -		· · ·	
· · · · ·	đi se			. ,

e e e e

· 5	(Co)			
Sample ID	Oore Wf (mg)	Au. (mg)	Ag (Troy 62/ton)	Au (Trug oz/ton)
3417	+ 6,62	0.264	12,712	0, 528
~ / / /	* 6.38	0,002	12,596	0,164
	* 17,31	0.515	33,59	(,030
3420	/0,8	0,2	21,20	0;4
	/1.1	0.2	21,80	0,4
	14,0	0.3	27.40	0.6
3422	8.8	۵.2	17.2	0.4
	12.5	0,1	24,8	0,2
	11, 8	0,/	23.4	0.2

 \mathcal{C}

* With these samples only the fire assay prills were sent to D.A. Sheh of Copper State Analytical in Thesm, AZ. In analysis by atmic adsorption, The prills were dissolved, the solution analysed for An, and the Ay determined by Subtraction. Prior to analysis. Mr. Shak reweighed the prills on his talance, which in the <0.1 mg range is now accurate them ours.

:

STATE OF ARIZONA



DEPARTMENT OF MINES AND MINERAL RESOURCES -

Mineral Building • State Fairgrounds • Phoenix, Arizona 85007

(602) 255-3791

March 22, 1989

Mr. Ken Broadhead Research Supervisor U S Bureau of Mines Reno Research Center 1605 Evans Avenue Reno, Nevada 89512-2295

Dear Ken,

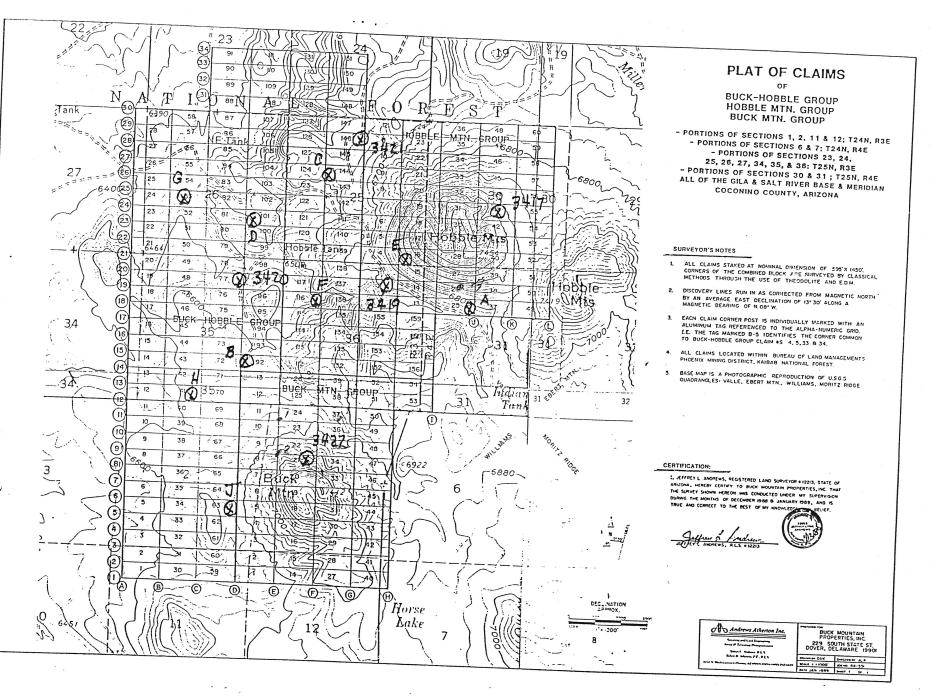
Just a quick note to convey a copy of the assay report on the samples that I took March 8, 1989 at the Buck Mountain Mining Project. Each one was taken near the respective sample taken by the Forest Service and are listed in the order in which they were taken. The split I gave you is from the last one (No. 28062). This was the ant hill.

I don't mean to second guess my own assayer but I will retain the rejects for possible reassay.

Hope your weather has been as nice as ours.

Sincerely,

Richard R. Beard Mining Engineer



Locations are approximate



ALVIN C. JOHNSON, JR., PH.D. EXPLORATION GEOCHEMIST 1707 EAST WEBER DRIVE, SUITE B TEMPE, ARIZONA 85281

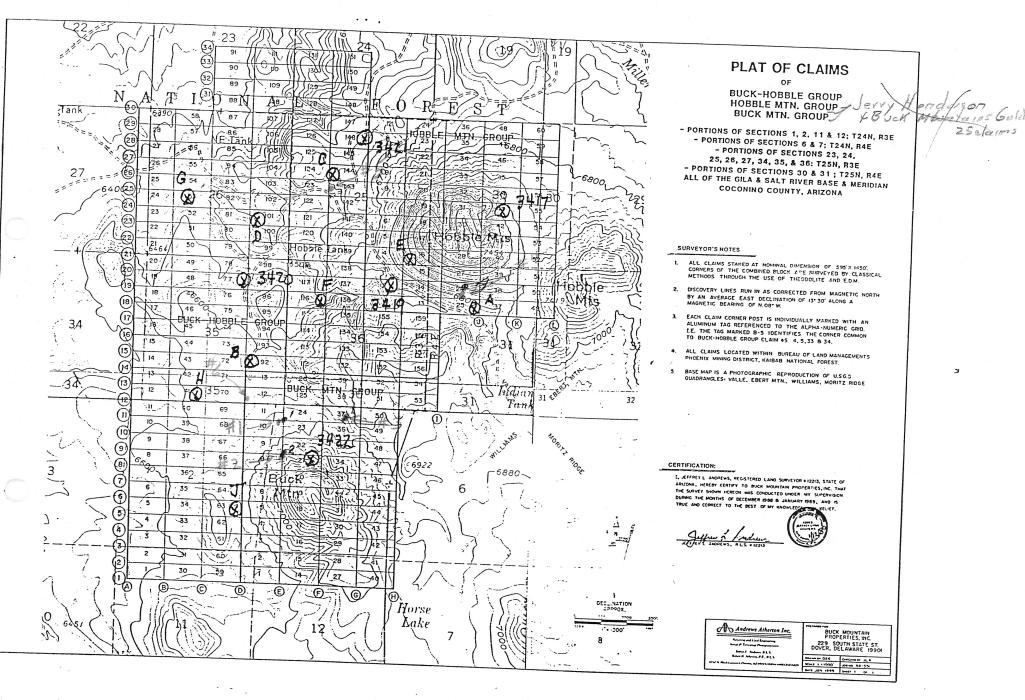
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Mix thoroughly and replace in the same clay crucible.

- 9. Fuse at 1000°C for 45 to 60 minutes (or until the fusion becomes quiescent).
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 - Place lead button in scorifying dish and add approximately 5 gram of assay lead. Cover the lead with borax and place in furnace at 1000°C. Scorify the lead until approximately one-half of the original lead remains. This step will take approximately 1.5 hours.
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Locations are approximate

IRON KING ASSAY INC.

Page 1

14-Mar-89

LAB JOB #:	MSC03820	
Client name:	AZ Dept. of Mines and Minera	lsNo. Samples: 6 Date Received: 03-08-89
Billing address:	Mineral Building Fairgrounds	Submitted by: R. Beard
Phone number:	Phoenix, AZ 85007 255-3791	INVOICE ATTACHED

ANALYTICAL REPORT

Client MSC03820	ID	Lab ID		Fire Assa Au oz/ton	y Ag oz/ton
	28057	3-8-	1	0.011	<.10
	28058	3-8-	2	0.006	<.10
	28059	3-8-	3	<.001	<.10
	28060	3-8-	4	0.001	<.10
	28061	3-8-	5	<.001	<.10
	28062	3-8-	6	0.001	<.10



A Long K MAR 2 3 1989

Dick

U.S. DEPARTMENT OF AGRICULTU	jre Date 6/6/89
REFERENCE SLIP	16/89
TO ·	
	NOTE AND RETURN
	PER PHONE CALL
AS REQUESTED	
	REPLY FOR SIGNATURE OF
	SEE ME
NOTE AND FILE	
	for your information
are the results	
1 11	ider material from
	properties on The
	Ohn Gubierrez has
0.	st Service minerals
position on proper	means of disposal of
this material. Th	Le Kaibab Forest intends
	esuct with the claimants
AZ Zone	Office
US Forest	Service
U.S.G.P.O.: 1985 - 526-216	FORM AD-514 (8-64)



United States Department of Agriculture Forest Service Arizona Zone Office 2324 E. McDowell Road Phoenix, AZ 85005 602 225-5261

Reply To: 2810

Date: June 1, 1989

Subject: Buck Mountain Properties

To: Dennis Lund, Kaibab NF

The assay results of the Buck Mountain sampling project have finally been received at the Arizona Zone Office in Phoenix. Briefly, the assays are results of an agreement between the Kaibab National Forest and Buck Mountain Properties to jointly sample and analyze volcanic materials for precious metals reportedly occurring on mining claims held by the latter. Prior to the sampling project, Buck Mountain Properties through their agent, Jefferey Andrews, agreed that the outcome of the sampling would determine what the company's next step would be towards development of the claims. The Forest Service' position is that the volcanic cinder material is a salable mineral and disposal is discretionary under 228 Subpart C regulations. Prior to the sampling effort, all parties understood that if assaying did not confirm the company's contention (.3 oz/ton gold and 30 oz/ton silver) of the presence of high gold and silver values, the Forest Service would insist upon the material's commonality. However, the Forest Service might consider sale of a limited quantity of similar volcanic material though not necessarily within the Buck Mountain claim block area.

As you might recall, the alleged valuable volcanic material has been previously sampled by Al Johnson, a consultant working for Buck Mountain. In order to duplicate earlier results, Johnson guided the recent sampling project to areas that were representative of the entire "deposit" and attempted to point out areas of the highest gold and silver values. Bulk samples were obtained from six locations within the claim block. These samples were later split using a standard rock/soil splitter and representative portions were distributed among the interested parties. Representatives from the U.S. Bureau of Mines, Arizona State Land Department, Al Johnson and the U.S. Forest Service each took a sample split to analyze for gold and silver content. A representative from the Arizona Department of Mines and Mineral Resources was also present but took separate samples of the volcanic material from areas nearby the bulk sample locations.

It was agreed that all samples would be sent to different laboratories in order to cross-check results by independent assayers using industry accepted fire assay procedures. Additionally, the Bureau of Mines agreed to assess Al Johnson's non-traditional assay procedures upon which Buck Mountain has relied exclusively to prove the presence of economic gold and silver.

The results of the sampling effort has been tabulated below. Unless otherwise specified, all samples were analyzed using standard industry accepted methods of fire assay:

Sample Number

Assaver	BM-1	BM-2	BM-3	BM-4	BM-5	BM-6
BuMINES						
Test ∦1	AU TR	TR	.04	.02	.01	TR
	AG .1	Nil	.20	.1	N11	Nil
#2	AU N11	N11	.03	N11	Ni]	N11
	AG .2	N11	.2	N11	.1	N11
<i>#</i> 3		(soil on (soil on L				soil only) soil only) soil only)
#4	AU Nil	TR	.04	TR	TR	TR
	AG .1	.1	.2	.1	.3	.3
** #5	AU N11	TR	.04	TR	TR	TR
	AG .1	.1	.4	.2	•4	.4
B-CLEGG	AU <.002	<.002	.029	<.002	<.002	<.002
	AG <.02	<.02	.24	<.02	<.02	<.02
IRON KING +	AU .011	.006	<.001	<.001	<.001	<.001
	AG <.10	<.10	<.10	<.10	<.10	<.10
SKYLINE	AU <.005	<.005	.13	<.005	<.005	<.005
	AG <.01	<.01	1.35	<.01	<.01	<.01

** Denotes assay results using Dr. Alvin Johnson's modified assay procedures. + Denotes samples taken adjacent to but different from split sample. TR = Trace or less than .005 oz/ton.

As can be seen, out of a total of the eighty-eight (88) assays processed, only Sample BM-3 reported low but anomalous values for gold and silver. The fact that all assayers but one found anomalous values in this sample tends to verify subeconomic amounts of precious metals are occurring. Skyline Laboratories reported a significantly higher precious metal content for BM-3, but this value was not duplicated by other assayers and thus is suspect. The remaining samples showed very low values, a trace of precious metals or minute quantities below the detection limitations of the equipment. The Bureau of Mines also conducted assay work for the platinum group metals using inductively coupled plasma analysis. Platinum, paladium and rhodium all recorded <.002 oz/ton readings. This extremely low figure is also below detection limitations of the equipment. Since the sample locations were agreed to be representative of the volcanic material contained within the entire claim block, the results support the Forest Service position that Buck Mountain mining claims contain common variety cinder-like materials.

As previously stated, Buck Mountain relied solely upon Al Johnson's earlier assaying work of the mining claims using a non-industry recognized method. It was necessary to assess Johnson's assay procedures in light of the escalated precious metal values (refer to Attachments) he found that were unlike anything previously known to be contained in similar volcanic material. Bureau of Mines Test #4 and #5 (above) compare the same sample using standard fire assay methods and Johnson's method. As can be seen, the samples check within acceptable limits. Bureau of Mines Research Supervisor, Ken Broadhead, states in his letter of May 18, 1989:

"...values obtained by using Dr. Johnson's procedure showed no difference from those obtained from normal fire assay techniques. The results of our work showed no large amounts of either gold or silver..."

Furthermore, Broadhead states that the new ... "procedure is unduly long, time consuming and completely unnecessary".

Similarly, Bondar-Clegg, an internationally recognized leader in assaying was asked for their assessment of Dr. Johnson's assay procedures. In a letter dated March 28, 1989, Greg Sprenger, fire assay lab manager states:

"...Bondar-Clegg feels that this method offers no significant improvement over stardard procedures routinely used in the assay industry. In fact, it is our opinion that the additional manipulations outlined in Dr. Johnson's procedure may increase the possibility for error in the analytical result."

Given the above data results and without further substantive evidence that economic gold and silver are contained in the volcanic material within the Buck Mountain claim group, it is my opinion that the material is common in nature. Therefore, disposal is discretionary and subject to regulations at 36 CFR 228 Subpart C. Buck Mountain Properties are not precluded from further exploration activities on their validly located mining claims even though the claims contain common volcanic materials. In the interest of fairness, I would recommend that the Forest sit down with the proponents to go over the assaying results and to discuss options available to them.

JOHN GUTIERREZ Mining Engineer

Attachments (3)

BC: Roger Masion, R.O. CC. Ken Broadhead; NV BUMines Dick Beard; AZ Dept Mines & Min Besources Gary Stusher; AZ Dept of State Land, Minerals Division



United States Department of the Interior

BUREAU OF MINES

RENO RESEARCH CENTER

1605 EVANS AVENUE RENO, NEVADA 89512-2295

May 18, 1989

John Gutierrez U.S. Forest Service 2324 East McDowell Road Phoenix, Arizona 85006-2497

Dear John,

Enclosed are the analytical reports on the rock and soil samples taken March 7, 1989, at the so called Buck Mountain Mining Project. The analyses are divided into two separate types, one to generally categorize and identify the material, and the other to quantify the precious metals content. The categorization tests as shown in exhibits 5, 6, and 7, show the volcanic cinders and soil to be primarily quartz, pyroxene and feldspars. There were no concentrations of elements present which would indicate that the samples would present any particular problems to standard fire assay techniques.

Exhibits 1 and 2 are fire assay results on my original splits for gold and silver. Exhibit 2 also shows the results following the procedure set forth by Alvin Johnson. A copy of this procedure is also enclosed as exhibit 8. Exhibit 3 shows regular fire assay results on your portions of material which I had re-requested from you and which had been forwarded to me by Bondar-Clegg. Exhibit 4 shows the results from a combined fire assay/ICP analysis for the platinum group metals.

Comments on exhibit 8: Dr. Johnson's assay procedure.

General: The procedure is unduly long, time consuming and completely unnecessary.

Specific: Steps 2 through 4: There is no reason for adding and sintering with sodium peroxide. There are no minerals present, or carbon or sulfur, which would require such an oxidizing agent.

Steps 5 through 7: No reason for the addition and sintering with PbO (litharge). I can not see the rational for such a step, particulary after the preceeding peroxide fusion.

J. Gutierrez

Steps 8 through 13. Scorification. There is no reason for the use of the excess flour to produce a large lead button which must be later scorified. Scorification of lead buttons is used primarily to either reduce the size of the lead buttons or to remove impurities such as copper and nickel from the buttons so that they will cupel properly. The elements however are not present in this particular material. Therefore, the whole exercise of excess flour and scorification serves no useful purpose. However, it may be detrimental to the procedure since the use of excess flour creates a lead deficient slag which may cause problems.

C. Cupellation 2. The door cannot be completely closed so as to exclude air since the air or actually oxygen is a necessary ingredient in the process. Without oxygen the molten lead will not form litharge (PbO) which will be adsorbed into the cupel--the purpose of the process. The cupellation temperature of 1,000° C is excessive and will lead to silver losses.

Generally speaking values obtained by using Dr. Johnson's procedure showed no difference from those obtained from normal fire assay techniques. The results of our work showed no large amounts of either gold or silver. Sample BM-4, however, did show low but anomalous amounts of gold.

Sincerely,

Kenneth G. Broadhead

Research Supervisor

Enclosures

cc: Dr. Al Johnson 1707 East Weber Drive Suite 8 Tempe, Arizona 85281

> Dick Beard Department of Mines and Mineral Resources Mineral Building-State Fair Grounds Phoenix, Arizona 85007

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March 28, 1989

John Gutierrez USDA, Forest Service Arizona Zone Office 2324 E McDowell Road Phoenix, AZ 85006

Dear Mr. Gutierrez,

Having reviewed the procedure by Dr. Alvin Johnson, "PROCEDURE FOR THE ANALYTICAL TESTING OF AU AND AG IN GEOLOGIC MATERIALS," which you enclosed with your recent submittal, Bondar-Clegg feels that this method offers no significant improvement over standard procedures routinely used in the assay industry. In fact, it is our opinion that the additional manipulations outlined in Dr. Johnson's procedure may increase the possibility for error in the analytical result.

Sincerely,

Bondar Cleac

Bondar-Clegg, Inc. 12980 West Cedar Dr. Lakewood, Colorado 80228 (303) 989-1404

QUETE ANALAA		PER STATE ANALYT AL LAB, INC. ARIZONA STATE CERTIFIED LAB NO. 0078 710 E. EVANS • TUCSON, AZ 85713 PH. (602) 884-5811 PH. (602) 797-0788			D.A. SHAH REGISTERED ASSAYER AZ REG. 8 4444 IN BUSINESS SINCE 1966	
SERVICES	7					
Mr. Alvin Johnso 1707 East Weber Tempe, Arizona 8	Drive, S	uite #8	. Re Sa			· · · · · · · · · · · · · · · · · · ·
Project:-William	m, AZ	•			* v	
Elements Units	Au OPT	Ag OPT		8222333	 	
0.8mg - BM-4 1.6mg - BM-2 2.2mg - BM-1 20.6mg - BM-5 24.4mg - BM-3 46.5mg - BM-1 88.3mg - BM-6	0.022 0.018 0.024 0.016 0.90 0.016 0.012	1.34 2.96 4.60 37.58 46.48 91.90 173.32				

NOTE:-Calculations are based on 15gm wt of sample. Client provided the dore from fire assay.

